Original Article

# Flavone constituents of *Phlomis bruguieri* Desf. with cytotoxic activity against MCF-7 breast cancer cells

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## Abstract

Phlomis bruguieri (P. bruguieri) is a large genus in the Lamiaceae family, with a wide variety distributed in Euro-Asia, Central Asia, Iran, and China. Phlomis flowers have been used as herbal tea for gastrointestinal disturbances, protection of liver and cardiovascular systems. The aim of this study was to analyse phytochemical of flavonoid constituents in semi polar fraction of *P. bruguieri*. Methanol extract of plant material (4 kg) yielded 361 g dark green concentrated extract gum. After preliminary fractionation by normal column chromatography on silica gel, Fr. 2 eluted with chloroform: methanol (90:10) selected as semi polar fraction and was more purified using different chromatography columns on silica gel, polyamide SC6 and Sephadex LH-20 adsorbents. Finally one new and three known flavonoids (1-4) were characterized in semi polar fraction. Isolated structures were identified using <sup>1</sup>H-NMR, <sup>13</sup>C-NMR, <sup>31</sup>P-NMR, HSQC, HMBC, negative ESI mass, and UV spectra using different shift reagents. Using standard MTT assay, cytotoxicity of isolated new compound was done against michigan cancer foundation-7 (MCF-7) breast cancer cells. Phytochemical analysis of *P. bruguieri* resulted in identification of one new 4'-methoxy-luteolin-7-phosphate and three known flavones including luteolin, apigenin, and tricin for the first time in this plant. In MTT cytotoxicity test, 4'-methoxy-luteolin-7-phosphate showed cytotoxicity with IC<sub>50</sub> value of 43.65 ± 8.56 μM agasint MCF-7 breast cancer cells.

*Keywords:* Breast cancer; Cytotoxicity; Flavone; MCF-7; Phlomis bruguieri.

# INTRODUCTION

The genus *Phlomis* from the Lamiaceae family are herbaceous, sub-shrub and rarely shrub with perennial growth. They are represented by 100 species from which 17-19 species are grown wild in Iran and specially in Zagros region (1,2). The genus is distributed mostly in world in Eur-Asia, Northwest of Africa, Turkey, Iran, central Asia, and China (3). In the case of secondary metabolites, the genus is rich in iridoids, diterpenes, flavonoids, phenylpropanoid glycosides, and phenylethanoid glycosides (4-7). Phlomis species are used in traditional medicine as analgesic, diuretic, emetic, and emmenagogue (8). Plant flowers are used in folk medicine as herbal tea for gastrointestinal problems and protection of liver and cardiovascular systems (9).

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Phlomis bruguieri (P. bruguieri) is one of the species distributed in flora of Iran, Turkey and Iraq (1). Previous studies on this plant have reported its antioxidant, antimicrobial Streptococcus sanguis against Staphylococcus and α-amylase aureus inhibitory activities (10-12). In the recent study, cytotoxicity of six Phlomis species against different human cancer cell lines was studied of which P. bruguieri showed interesting cytotoxic activity against MCF7 breast cancer cell line (13).

The main constituents of essential oils included germacrene D, apiole, and myristicin (9).

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In another study by high performance liquid chromatography electrospray ionization tandem mass spectrometry (HPLC-ESI/MS) analysis screening on aerial parts of bruguieri, phenylethanoid glycosides isoverbascoside, including verbascoside, leucosceptosides A and martynoside were reported and quantified (14). In a study conducted by Kaaji et al, it was shown that Phlomis species have a wide variation in flavonoid contents including flavones. isoflavonones, and chalcones. Therefore, the paper in hand aimed to isolate and characterize flavonoid constituents in P. bruguieri in addition to cytotoxic evaluation of isolated new compound against breast cancer cells.

# MATERIALS AND METHODS

# General procedures

Nuclear magnetic resonance analysis (NMR) was done on a Bruker AV400 spectrometer (Billerica, MA, USA) using deuterated dimethyl sulfoxide (DMSO-d<sub>6</sub>) as the solvent at 400 MHz for proton nuclear magnetic resonance (1H-NMR), 100 MHz for carbon-13 (13C)-NMR, and 31 MHZ for phosphorus-31 (<sup>31</sup>P)-NMR. Mass spectra (electrospray ionization (ESI)) were performed on Shimadzu 2010EV liquid chromatographyspectrometry (LC-MS) mass system (Shimadzu, Japan), and are reported in m/z. Thin layer chromatography (TLC) was done on Merck TLC silica gel allufoils (Germany) and visualized with 1% natural product reagent (2-aminoethyl diphenylborinate) and 1% ceric sulfate solution in 10% sulfuric acid followed by heating by hair dryer for about 2 min. Column chromatographies were done on silica gel (60-200 tm, Merck, Germany), polyamide SC6 Germany), Sephadex-LH (Roth, (Pharmacia fine chemicals, Uppsala, Sweden). Sephadex LH-20 was swelled in hexane: methanol: acetone (30:60:10) in a glass column 45 cm in length and 2 cm i.d., fitted with teflon stopcock and rinsed with the same solvent system with a flow of 3 cm/h.

#### Plant material

Aerial parts of *P. bruguieri* Desf. (Lamiaceae) was collected from Zagros region

in Kermanshah, west of Iran. Plant material was identified by taxonomist according to the voucher specimen (2182) deposited in the herbarium of biology department, Faculty of Science, University of Isfahan, Isfahan, I.R. Iran.

#### Extraction and isolation

The air-dried powder of the plant material (4 Kg) was extracted trice with methanol  $(20 L \times 3)$  at room temperature for one week. Combined extracts were concentrated to a dark brownish gum (361 g) by rotary evaporator attached to vacuum pump (60 mbar) at 40 °C. The extract was suspended in water and defatted in a separating funnel by hexane. Defatted aqueous phase was concentrated (80 g) and adsorbed on pre-adsorbent (Celite) in equal weight and then applied on silica-gel column chromatography (400 g) for preliminary fractionation. Liquid chromate-graphy using solvent systems of hexane:chloroform (70:30), chloroform: methanol (90:10), and methanol (100%) yielded three fractions (Frs.1-3). Fr.1 eluted with hexane:chloroform (70:30)containing fatty acids and Fr.3, eluted with methanol (100%) containing polar glycoside secondary metabolites were put aside to evaluate later. Semi-polar fraction, Fr. 2 eluted chloroform:methanol (90:10)with selected and subjected silica on gel column chromatography (hexane:acetone, 90:10, 85:15, 80:20, 70:30, and 50:50) and yielded 5 sub-fractions (Frs. 2A-2E). After taking preliminary <sup>1</sup>H-NMR spectra of different fractions, Fr. 2C, Fr. 2D, and Fr. 2E showing flavonoid profiles were selected and subjected to more purification on polyamide SC6 column chromatography using stepwise gradient of chloroform:methanol (98:2, 96:4, 94:6, 92:8, 90:10, and 80:20). Based on TLC profiles visualized by natural product reagent, sub-fractions with vellowish spots in TLC was indicative of flavonoid content and were further purified by size exclusion chromatography on a Sephadex LH-20 column  $(2 \times 45)$  using hexane:methanol:acetone (30:60:10) as the mobile phase. Finally, Fr. 2C6a (12.1 mg), Fr. 2D3 (8.3 mg), Fr. 2D5b (6.8 mg), and Fr. 2E7a (11.2 mg) were obtained as pure compounds called 1-4 and submitted for identification to NMR, mass spectra (15).

# Cytotoxicity assay

Because of anti-estrogenic properties of flavones, isolated new flavonoid was subjected to MTT assay on estrogen sensitive MCF-7 breast cancer cells. Pasteur Institute of Iran (Tehran, I.R. Iran) provided MCF-7 cell line which was grown adherently in RPMI-1640 with 10% fatal calf serum (FCS), 100 units/mL penicillin and 100 μg/mL streptomycin at 37 °C in 5% CO2. Cells were seeded at 5000 cells per well in 5% CO2 at 37 °C in RPMI with 10% FBS, 100 U/mL penicillin and 100 μg/mL streptomycin, in 96-well plates. Compound 2 (2 mg) was dissolved in DMSO (1 mL) for preparation of stock sample (5.27 mM).

Different concentrations of compound were prepared by serial dilution method using RPMI as diluent. After 24 h incubation, cells exposed with concentrations of 0.1, 1, 10, 50, and 100 µM of tested compound for 48 h. Doxorubicin as positive control was used at concentrations 0.001, 0.01, 0.1, 1, 10, and  $20 \mu M$ . MTT was added to the wells and incubated for another 4 h. Experiment was done in triplicate and the absorbance was read by the microplate reader (Bio-Rad, Hercules, CA, USA) at 570 nm. Cell viability percentages were calculated by the following equation and were expressed as percent of control cells which were not treated (16,17).

Fig. 1. Flavones isolated from *Phlomis bruguieri*.

Cell viability
$$= \frac{Optical\ density\ mean\ of\ treated\ cells}{Optical\ density\ mean\ of\ control\ cells} \times 100$$

# Statistical analysis

Biological data are reported as mean  $\pm$  SD and the IC<sub>50</sub> values were calculated using Excel worksheet processor. T-test with two samples assuming equal variances as well as Dunnet test one way ANOVA were used for analysis of MTT cytotoxicity results and P < 0.05 was considered to indicate statistically significant differences.

#### **RESULTS**

Methanol extract of plant material (4 Kg) yielded 361 g dark green gum. After preliminary fractionation by silica gel column, Fr. 2 eluted with chloroform:methanol (90:10) was selected and more purified using different chromatography adsorbents including silica gel, polyamide SC6 and Sephadex LH-20. Finally one new and three known flavonoids (1-4) were characterized in semi polar fraction (Fig. 1). Isolated structures were identified <sup>13</sup>C-NMR, <sup>31</sup>P-NMR, <sup>1</sup>H-NMR, using single quantum correlation heteronuclear (HSQC), heteronuclear multiple correlation (HMBC), negative ESI-mass, and UV shift reagents as well as comparing with literature data (Fig. 2).

Fig. 2. Electron ionization mass fragmentation pattern of 4'-methoxy-luteolin-7-phosphate. Cleavage of ring C by retro-Diels-Alder (RDA) mechanism led to A (m/z 231) and B (m/z 117) ions (16).

# Spectral data of isolated compounds

Compound **1**: Pale yellow powder, <sup>1</sup>H-NMR in DMSO-d<sub>6</sub> (400 MHz) ppm  $\delta$  5.39 (1H, d, J = 2.0 Hz, H-8), 5.62 (1H, d, J = 2.0 Hz, H-6), 5.72 (1H, s, H-3), 6.10 (1H, d, J = 8. 8 Hz, H-5'), 6.56 (1H, overlapped, H-2'), 6.58 (1H, dd, J = 2.0, 8.8, H-6'); <sup>13</sup>C-NMR (100 MHz, DMSO-d<sub>6</sub>)  $\delta$ <sub>C</sub>: 181.6 (C4), 164.1 (C2), 163.9 (C7), 161.4 (C5), 157.2 (C9), 149.7 (C4'), 145.7 (C3'), 121.4 (C1'), 118.9 (C6), 116.0 (C5'), 113.3 (C2'), 103.6 (C10), 102.8 (C3), 98.8 (C6), 93.8 (C8), 55.9 (4'-OMe). Negative ESI mass (m/z) 285 [M-H]<sup>-</sup>.

Compound **2**: Pale yellow powder,  ${}^{1}H$ -NMR in DMSO-d<sub>6</sub> (400 MHz) ppm  $\delta$  3.95 (1H, s, 4'-OMe), 6.26 (1H, d, J = 2.0 Hz, H-8), 6.58 (1H, d, J = 2.0 Hz, H-6), 6.96 (1H, s, H-3), 6.99 (1H, d, J = 8. 8 Hz, H-5'), 7.61 (1H, overlapped, H-2'), 7.63 (1H, overlapped, H-6');  ${}^{13}C$ -NMR (100 MHz, DMSO-d<sub>6</sub>)  $\delta_{C}$ : 181.8 (C4), 164.1 (C2), 163.6 (C7), 161.4 (C5), 157.3 (C9), 150.7 (C4'), 148.0 (C3'), 121.5 (C1'), 120.3 (C6'), 115.7 (C5'), 110.1 (C2'), 103.7 (C10), 103.2 (C3), 98.8 (C6), 94.0 (C8).  ${}^{31}P$ -NMR in DMSO-d<sub>6</sub> (162 MHz) ppm  $\delta$  0.18 (C7-H<sub>2</sub>PO<sub>4</sub>). Negative ESI mass (m/z): 379 [M-H]<sup>-</sup>, 343 [379-2H<sub>2</sub>O]<sup>-</sup>, 299 [379-H<sub>2</sub>PO<sub>3</sub>]<sup>-</sup>, 271[299-CO], 257, 231, 151, 117.

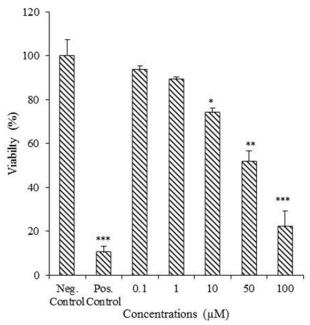
Compound 3: Pale yellow powder, <sup>1</sup>H-NMR in pyridine-d<sub>6</sub> (400 MHz) ppm: 6.10

(1H, d, J = 2.0 Hz, H-8), 6.36 (1H, d, J = 2.0 Hz, H-6), 6.49 (1H, s, H-3), 6.83 (2H, d, J = 8.8 Hz, H-6',2') and 7.75 (2H, d, J = 8.8 Hz, H-5',3'); <sup>13</sup>C-NMR (100 MHz, pyridine-d<sub>6</sub>)  $\delta$ C: 183.2 (C4), 166.4 (C2), 163.8 (C7), 163.3 (C5), 163.2 (C4'), 159.0 (C9), 129.4 (C5', C3'), 117.3 (C6', C2'), 122.8 (C1'), 106.5 (C10), 104.4 (C3), 100.5 (C6), 95.3 (C8). Negative ESI mass (m/z): 269 [M-H]<sup>-</sup>.

Compound 4:  ${}^{1}$ H-NMR (400 MHz, DMSO-d<sub>6</sub>) ppm 4.04 (s, 6H, 3'-OMe, 5'-OMe), 6.30 (1H, d, J = 2.0 Hz, H-6), 6.55 (1H, d, J = 2.0 Hz, H-8), 6.69 (1H, s, H-3), 7.32 (2H, s, H-6',2');  ${}^{13}$ C-NMR (100 MHz, pyridine-d<sub>6</sub>)  $\delta$ C: 183.7 (C4), 166.1 (C2), 165.6 (C7), 163.3 (C4'), 163.3 (C5'), 159.2 (C9), 150.1 (C5', C3'), 141.7 (C1'), 123.4 (C10), 106.2 (C6', C2'), 105.5 (C3), 100.7 (C6), 96.1 (C8), 58.2 (3'-OMe, 5'-OMe). Negative ESI mass (m/z): 329 [M-H]<sup>-</sup>.

## Result of cytotoxicity assay

The MTT assay was done on compound 2 as a new compound to check its cytotoxicity against MCF-7 breast cancer cells (Fig. 3). MTT results showed moderate cytotoxicity with IC<sub>50</sub> value of  $43.65 \pm 8.56$   $\mu M$  while doxorubicin as standard drug showed IC<sub>50</sub> value of  $0.32 \pm 0.08$   $\mu M$ .



**Fig. 3**. Cytotoxicity effect of 4'-methoxy-luteolin-7-phosphate against MCF-7 breast cancer cells. Cells were treated with different concentrations of tested compound (0.1, 1, 10, 50, and 100  $\mu$ M) in three replicates. Doxorubicin (10  $\mu$ M), and vehicle were used as positive and negative controls, respectively. (\* P < 0.5; \*\* P < 0.01; \*\*\* P < 0.001 vs control).

#### DISCUSSION

Compound 1 was isolated as a pale yellowish solid with positive reaction to FeCl<sub>3</sub> flavonoid natural product reagent. Molecular formula was determined <sup>13</sup>C-NMR data (broad band and distortionless enhancement by polarization transfer (DEPT)) and negative ESI/MS m/z 285 [M-H] as  $C_{15}H_{10}O_{6}$ . The spectrum UVshowed absorption maxima at 264 and 350 nm characteristic of flavones (16). The <sup>1</sup>H-NMR spectrum displayed two meta coupled doublets at  $\delta_{\rm H}$  6.26 (1H, d, J = 1.6 Hz) and 6.45 (1H, d, J = 1.6 Hz) attributed to H-6 and H-8, a singlet proton at  $\delta_H$  6.51 (1H, s, H-3) two ortho coupled proton signals at  $\delta_{\rm H}$  6.97 (1H, d, J = 8.4 Hz) and 7.33 (1H, dd, J = 8.4, 2.0 Hz), corresponding to H-5' and H-6' as well as  $\delta$ 7.41 (1H, d, J = 2.0 Hz) corresponding to H-2' proton, indicated that compound 1 is 3',4',5,7 tetrahydroxy flavone known as luteolin in agreement with literature data and co-TLC with authenticated sample (18,19).

Compound **2** showed positive reaction to flavonoid natural product reagent.  $^{1}$ H-NMR spectra resonated at  $\delta_{\rm H}$  6.26 (1H, d, J = 2.0, H-6), 6.58 (1H, d, J = 2.0, H-8), 6.96 (1H, s, H-3), in addition to an ABX spin system at

 $\delta_{\rm H}$  6.99 (1H, d, J = 8.8 Hz), 7.63 (1H, dd, overlapped), and 7.62 (1H, bd, overlapped) corresponding to H-5', H-6', and H-2' showed closed similarities to those of compound 1 but an additional methoxy signal at  $\delta_H$  3.95 (3H, s) and an additional phosphate group at  $\delta_P$  0.18 ppm which made slight downfield shifts in aromatic protons resonances in ring A.  $^{2,3}J_{C-H}$ HMBC correlations of methoxy singual with  $\delta_{\rm H}$  3.95 (3H, s) and  $\delta_{\rm C}$  150.7 ppm confirmed the location of methoxy group at C-4' in ring B. Structural assignment of compound 2 was initially assumed to be the corresponding sulfate analog on the basis of the facts that sulphate esters of flavonoids were well reported in literature, but it was eventually identified by <sup>31</sup>P-NMR to be the corresponding phosphate ester. Presence of a phosphate group was confirmed through <sup>31</sup>P-NMR with proton decoupling at  $\delta_P$  0.18 ppm (Fig. 2). Negative ESI/MS showed m/z 379 as molecular ion [M-H] together with fragment ions of m/z 343 [379 - 2H<sub>2</sub>O], 299 [379 - $H_2PO_3$ ], and retro-Diels-Alder (RDA) fragments of 231, and 117 ions were indicative of a flavone with one phosphate on ring A and one methoxy and one hydroxyl group on ring B (Fig. 2). <sup>13</sup>C-NMR showed 16 decoupled signals including one methoxy at  $\delta_{\rm C}$  55.9 (4'-OMe), 6 olefin methines and 8 quaternary sp<sup>2</sup> carbons from which 6 were oxygenated (experimental section), and a carbonyl carbon ( $\delta_C$  181.8 ppm). The UV spectrum showed absorption maxima at 267 and 350 nm characteristic of flavones (16). Bathochromic shifts of band III at 386 nm in AlCl<sub>3</sub>/HCl was about 36 nm indicative of a free hydroxyl group at C-5 (16). Addition of NaOAc following with BH<sub>3</sub> did not change band I and III and was indicative of lack of free hydroxyl group at C-7 in ring A. Therefore, based on these data, the structure of compound 2 is proposed as 4'-methoxyluteolin-7-phosphate but needs more analytical analysis for excat structure.

Phosphate-transferring phosphatases made phospho-intermediates including phosphate and pyrophosphate esters secondary metabolite as biosynthetic intermediates like erythrose-4-phospate, inositol-phosphate, glyceraldehyde phosphate, and etc., which are hydrolyzed at the end of biosynthesis pathway. Therefore, phosphointermediates found in small amounts and a limited number of them are identified till now. In comparison with phosphate estrs, inorganic sulphates conjugated forms of flavonoids like apigenine-7-sulphate, and quercetin-3-sulphate are more reported in literature. In a study conducted by Harborn et al. they reviewed and reported distribution and structural variations of inorganic sulphate esters of flavonoids (20).

Compound 3 was isolated with a pale yellowish color and positive reaction to natural product reagent. The UV spectrum showed two intense absorption maxima at 266 and 339 nm characteristics of flavones. Based on NMR data and negative mode of ESI/MS m/z 269 [M-H] molecular formula was proposed as C<sub>15</sub>H<sub>10</sub>O<sub>5</sub>. The <sup>1</sup>H-NMR spectrum displayed two meta doublets at  $\delta$  6.10 (1H, d, J = 2.0Hz), and 6.36 (1H, d, J = 2.0 Hz) coupled to each other similar to those of H-8 and H-6 in flavones in addition to a singlet proton at 6.49 (H-3), and two *ortho*-coupled signals at  $\delta$ 6.83 (2H, d, J = 8.8 Hz) and 7.75 (2H, d, J = 8.8 Hz), describing to H-6',2' and H-5', 3' related to AA'BB' spin system indicated that compound 3 is 4',5,7-trihydroxyflavone or apigenin (19).

Compound 4 with pale color showed UV spectrum pattern of flavones. The <sup>1</sup>H-NMR spectrum displayed two coupled meta doublets at  $\delta$  6.30 (1H, d, J = 2.0 Hz), and 6.55 (1H, d, J = 2.0 Hz), and a singlet signal at 6.69 (1H, s), related to H-6, H-8, and H-3 similar to apigenin but different in ring B in AA' spin system at δ 7.32 (2H, s) instead of AA'BB' system ascribing to H-6',2' protons, as well as a singlet signal at 4.04 (6H, s) related to two symmetric methoxy groups located at C-5', and C-3'. Therefore, based on <sup>1</sup>H- and <sup>13</sup>C-NMR data (see experimental), ESInegative mass m/z 329 [M-H], and literature data it was identified as 4',5,7-trihydroxy-3', 5'-dimethoxyflavone or tricin (21).

These results were in agreement with previous report on presence of different classes of flavonoids in *Phlomis* species. In a study conducted by Aghakhani and coworkers using chromatography liquid tandem spectrometry as a fast but not very certain screening method, they proposed presence of different flavonoids including kaempferol, chrysoeriol, naringenin, and luteolin, 4-hydroxy-5,7-dimethoxy flavanone in P. bruguieri leaves (22).

In biological assay, compound 2 showed cytotoxic activity with IC<sub>50</sub> value  $43.65 \pm 8.56 \mu M$  against estrogen sensitive MCF-7 breast cancer cells. MTT result was in with agreement antiestrogenic antiproliferative activities of flavonoids against MCF-7 breast cancer cells (23-25). In a study done by Bail et al. they evaluated flavonoids for their cytotoxicity against estrogen-dependent (MCF-7) breast cancer cells (26). In their study, 7-methoxyflavanone 7,8-dihydroxyflavones showed estrogenic and cytotoxic properties. In another study on luteolin by Wang, luteolin inhibited MCF-7 cells proliferation which was induced pathway dependent estrogen IGF-1 receptor-α (23). It regulated also estrogen signaling and cell cycle pathway genes (26). another study on chrysin dihydroxyflavone) and its phosphate esters on HeLa cancer cells, the results showed both suppressed compounds HeLa cells proliferation by apoptotic induction. comparison between parent and its phosphate ester, chrysyl-7-phophate showed more cytotoxic properties which is probably due to its better solubility (27). Low aqueous solubility is one of the main problems of flavonoid applications leads to their poor oral bioavailability which could be improved by flavonoid phosphate esters and making their potassium or sodium salts (28).

# **CONCLUSION**

Phytochemical analysis of *P. bruguieri* resulted in identification of one new 4'-methoxy-luteolin-7-phosphate and three known flavones including luteolin, apigenin, and tricin for the first time in this plant. In MTT cytotoxicity test, 4'-methoxy-luteolin-7-phosphate showed moderae toxicity agasint MCF-7 breast cancer cells.

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